

Incubation study on the effect of soil moisture and rainfall on NH₃ emissions from different inhibited urea fertilizers

Introduction:

- NH₃ emissions are affected by soil and weather conditions after fertilizer application.
- Influence of soil water content likely not linear, low NH₃ emissions only under dry and very wet soil moisture conditions.
- With medium water content: fast dissolution of the fertilizer granule, fast hydrolysis of urea, low pH buffering due to low spatial distribution in the soil (see Fig 1), resulting in high emission potential.
- How do urease- and nitrification inhibitors interact on NH₃ emissions at different soil water contents and after rainfall?



Fig 1: Hygroscopic urea granule on dry soil starting to dissolve after a dewy night.

Materials & Methods:

- Incubation study in controlled environment → NH₃ emission potential

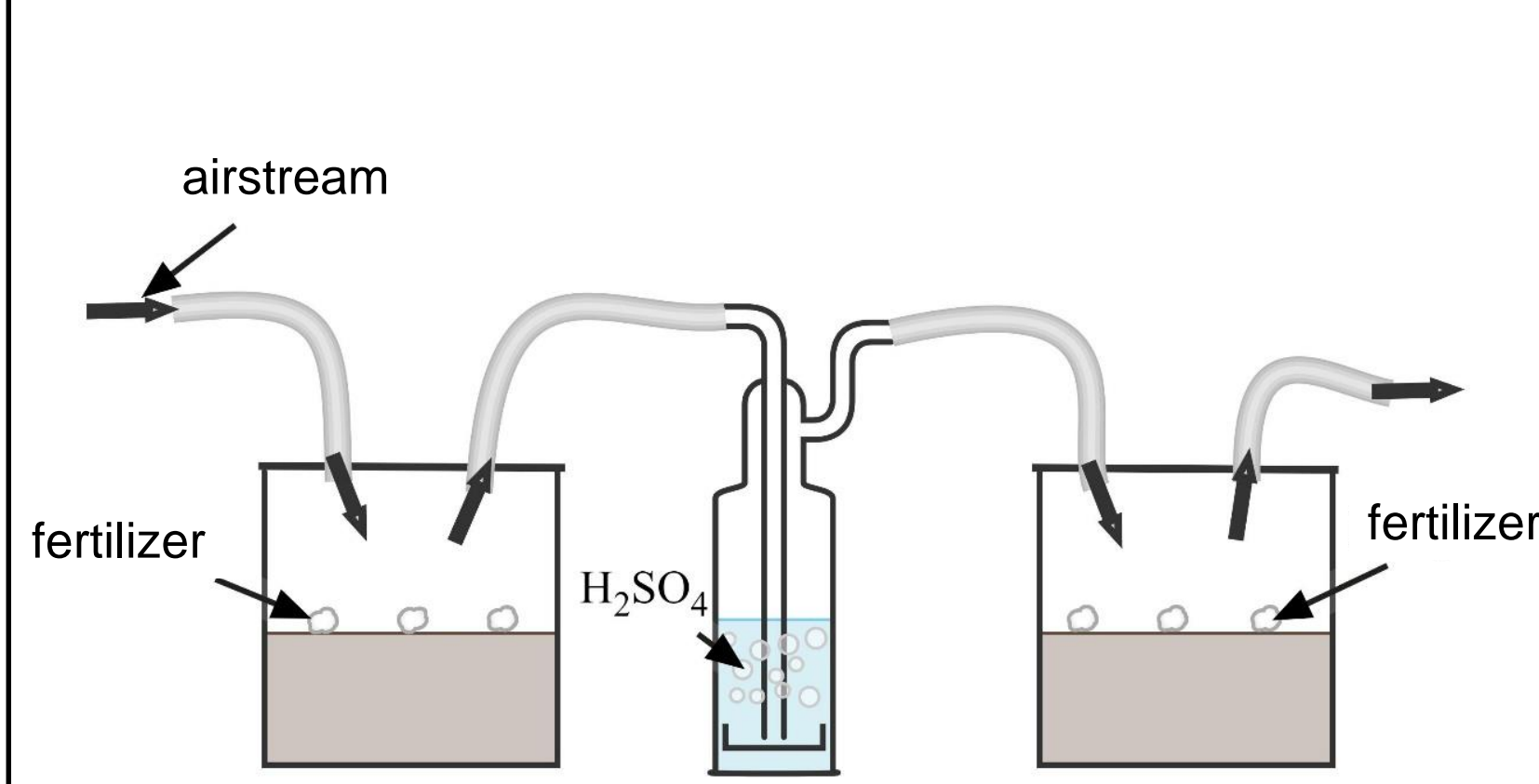


Fig 2: Experimental setup from main incubation glass (left), NH₃ scrubbing bottle containing H₂SO₄ (middle) and incubation glass for soil samples (right)

	Experimental run 1	Experimental run 2
Influencing factor	Soil water content	Rainfall after fertilization
Soil	Sandy loam	
Fertilizer rate	100 kg N ha ⁻¹	
Air exchange	1 headspace min ⁻¹	
Fertilizer treatments	Unfertilized control (N0), Urea (U), Urea + urease inhibitor (U+UI), Urea + urease inhibitor + nitrification inhibitor (U+UI+NI)	
Gravimetric water content	10, 17.5, and 25 %	17.5 %
Simulated rainfall	None	1, 5, 10 mm 4 days after fertilizer application

Results & Discussion:

Water content:

- Low water content resulted in low emissions (Fig 3b), because of the slow fertilizer granule dissolution. No NH₄⁺ accumulation due to simultaneous nitrification (not shown).
- Medium water content resulted in highest NH₃ emissions (Fig 3a), as NH₄⁺ release from urea hydrolysis was faster than NH₄⁺ oxidation during nitrification (not shown).
- High water content resulted in low emissions, likely due to a better diffusion of urea exploiting a higher soil volume, thus increasing soil pH buffer.

Fertilizer type:

- Urease inhibitor resulted in a slower NH₄⁺ release compared to none inhibited urea. A lower NH₃ peak occurred 4 days delayed.
- As supported by the N_{min} values (data not shown), the nitrification inhibitor resulted in an accumulation of NH₄⁺ thus enhancing NH₃ volatilization, compared to U+UI (Fig 3b).

Rain amount:

- The higher the rainfall, the lower the NH₃ emissions (Fig 3d).
- A combination of rainfall and inhibitor always decreased emissions (Fig 3d) compared to no rainfall (Fig 3b).
- At rainfall amounts ≥ 5 mm, NH₃ emission from U+UI treatment and N0 did not significantly differ (Fig 3d).
- Each rainfall amount (Fig 3d) significantly reduced NH₃ loss when compared to the treatment without rain (Fig 3b).

Conclusion

Soil moisture is a strong driver for NH₃ emissions after fertilizer application. Rainfall effectively decreases emissions. UI reduces emissions effectively, while UI+NI can increase emissions.

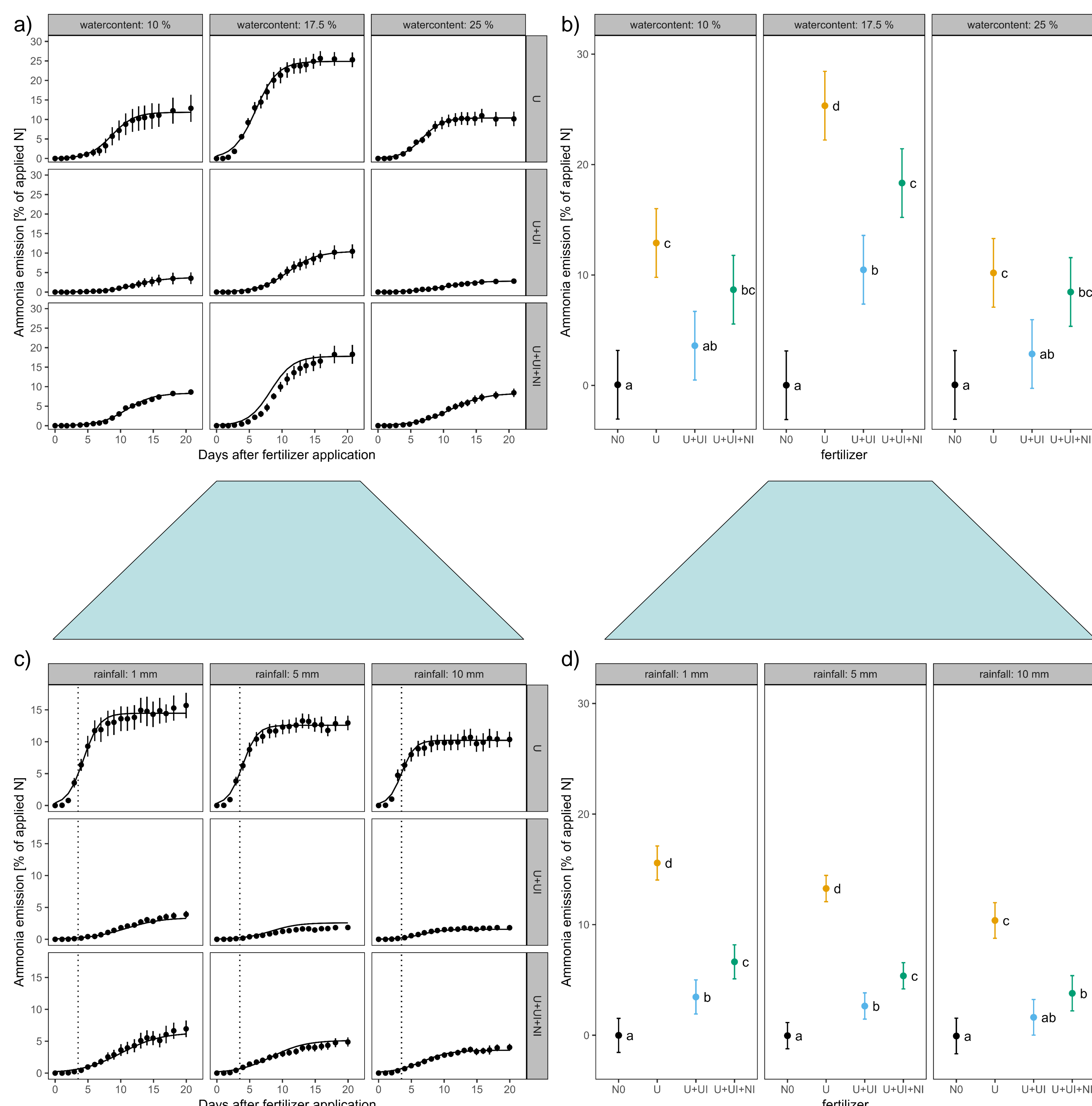


Fig 3: a)+c) Daily cumulative emissions (kg N ha⁻¹ day⁻¹, corresponding to % of applied N), as affected by fertilizer and gravimetric soil water content (a) or by fertilizer and rain amount (c). Error bars give standard deviation. b)+d) Cumulative emissions at last experimental day, as affected by fertilizer and soil water content (b) or by fertilizer and rain amount (d). Dots and error bars represent estimated means ± 95% confidence interval per group. For each water content and rain amount separately, means not sharing any letter are significantly different by the Tukey-test (p < 0.05).

